## Units of Measurement

## SI Units

| Seven Base <br> SI units | Derived units | Supplimentary <br> units |
| :---: | :---: | :---: |

$1 \mathrm{~A}^{\circ}$ (angstrom) $=10^{-10} \mathrm{~m}$ 1 fermi $=10^{-15} \mathrm{~m}$ $1 \mathrm{AU}=1.49 \times 10^{-11} \mathrm{~m}$ $1 \mathrm{ly}=9.46 \times 10^{-15} \mathrm{~m}$ 1 parsec $=3.08 \times 10^{-16} \mathrm{~m}$

## Units of Mass

Mass of atoms in a.m.u. 1 metric ton $=10^{3} \mathrm{~kg}$
1 solar mass $\approx 10^{30} \mathrm{~kg}$

Interstellar Distance


## Parallax method

Method to measure such interstellar distances from earth


## Dimension

A "Dimension" can be measured or derived.
The "Fundamental dimensions" (length, time, mass, temperature, amount) are distinct and are sufficient to define all the others.

Five Base Quantities

| Mass M | Length L | Time T | Electric Current A | Temperature K |
| :--- | :--- | :--- | ---: | :--- |
| Dimensional Analysis |  | Significant Figures |  |  |

- Show the relationship between

Accurately known digits plus first uncertain digit in a measurement
different system of units

- Implicitly tell how to derive a relation
- Provide a check on relation between quantities Rule 1: All non-zero digits are always significant

Rule 2: Zeros in between significant figures are always significant

Rule 3: Space holder zeros in numbers < 1 are never significant Rule 4: Zeros at the end of a number are only significant when a decimal is in the number

Errors in Measurement
Uncertainty in the measured values

## Systematic Errors

 Range of observed values Precision of measurement Minimizing Method By improving the structures of apparatusis reduced by subtracting the obtained reading fom the zero error Incorrec
calibration

## Systematic Errors Vs Random Errors

## Systematic Errors

Systematic error is the one that deviates from the true value of measurement by a fixed amount.

It remains constant or changes in a regular fashion in repeated measurements of the same quantity.

Caused by some flaw in the experimental apparatus or a flawed experimental design.

It can be eliminated using proper technique, calibrating equipment and employing standards.

## Combination of Errors

| $Z=A \pm B$ | $Z=A^{2}$ | $Z=A \cdot B$ or $A / B$ |
| :---: | :---: | :---: |
| $\Delta Z=\triangle A+\triangle B$ | $\frac{\Delta Z}{Z}=2 \frac{\Delta A}{A}$ | $\frac{\Delta Z}{Z}=\frac{\Delta A}{A}+\frac{\Delta B}{B}$ |

